

POTASSIUM ALCOHOLATES

Potasium t-Amylate [KTA] In Xylene 25%

a. 25 wt% Solution in Xylene

b. Density at 25°C Approx 0.9 gm/ml

1. OTHER NAMES

a] Potassium-tert-pentoxide in Xylene 25% b] Potassium-t-amylate in Xylene 25% c] Potassium-t-amoxide in Xylene 25% d] KTA in Xylene 25%

2. CAS NO.

a. 41233-93-6 for KTA b. 1330-20-7 for Xylene

3. FORMULA WEIGHT

126.2 gm/mole

4. TECHNICAL SPECIFICIATION

- a. Appearance: Yellow liquid
- b. Total alkalinity (%): 25-27
- c. Hydroxide content (%): 1 max
- d. KTA content (%): 24-26

5. SOLUBILITY

KTA is very soluble in toluene, Wylene, Cyclohexane, hexane, diglyme and tetrahydrofuran.

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6. STABILITY

Atmospheric moisture and carbon dioxide reacts readily with KTA to produce potassium hydroxide and potassium carbonate. Tertlary amyl alcohol is liberated from these reactions. It develops yellow to brown color solution after reacting with water. KTA solution should be stored in a cool place away from heat, sparks and flame.

7. PACKAGING

- a. Sample packing from 100 gms to 500 gms in glass bottle
- b. 180 kgs in 210 lit. steel drum
- c. Any other packing as per customer request

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8. SAMPLING INSTRUCTIONS

- a. The product is packed under dry nitrogen with positive pressure of nitrogen inside the drum.
- b. The quality of the product deteriorates very fast if exposed to atmosphere even for a brief period.
- c. While sampling, please ensure that the sample is taken out under dry nitrogen in a preweighed stoppered bottle and analysis is done immediately.
- d. After sampling, close the container securely after putting positive nitrogen pressure in the drum. This is very important so that the product does not deteriorate on storage.

9. SHIPPING INFORMATION

a. UN-2920, PG 1

b. Corrosive flammable liquid

10. PRODUCT PROPERTIES

- a. Very high purity
- b. Very strong base
- c. Low hydroxyl content
- d. Selective and specific in many organic reactions
- e. Stronger base than primary and secondary alcoholates
- f. Custom packaging available
- g. Any quantities in bulk

11. PRODUCT BENEFITS

- a. Strong hydrocarbon soluble base USED FOR:
- **b.** Deprotonations
- c. Base catalyzed reactions
- d. Elimination reactions
- e. Super base reaction with butyllithium
- f. Isomerization reaction

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